



Fig. 1. Heat stable muscle protein. (Magnification 850 \times)

tation rate of very elongated particles depends primarily on width and very little on length, the homogeneity suggested by a sharp sedimenting boundary may be illusory, and the results compatible with the assumption that the material consists of thread-like particles of uniform width, in various degrees of linear aggregation. Osmotic pressure measurements by the senior author's method⁷ yielded only erratic results, a fact usually observed with material that can be suspected of physical instability.

JACQUES BOURDILLON

*Division of Laboratories and Research, New York State Department of Health, WINFIELD H. BAKER
Albany, N.Y. (U.S.A.)*

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Synthesis of ribose by the rat*

The biosynthesis of ribose¹ has been postulated to occur (a) directly from hexose by loss of an end carbon and (b) from smaller units by condensation, *e.g.* $C_2 + C_3$. Data obtained in previous experiments² have been interpreted to indicate that in the chick, the synthesis of this pentose probably occurs by the latter mechanism, *possibly* by a condensation of glyceraldehyde-3-phosphate and an active C_2 unit (*e.g.* from fructose-6-phosphate³) catalyzed by transketolase. The technique employed was to isolate glycogen and ribose from the combined internal organs

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of chicks after the administration of ^{14}C -labeled precursors and to compare the patterns of ^{14}C in the two carbohydrates. If ribose were synthesized from hexose directly, the pattern of ^{14}C in the ribose should be similar to that of five consecutive carbons in the hexose. Dissimilar patterns would indicate the probable existence of another mechanism. The patterns of ^{14}C in the glycogen and ribose isolated from the tissues of the chicks were markedly different and it was concluded that the data were consistent with the hypothesis that the pentose arose by combination of C_2 and C_3 units. The applicability of this conclusion to other organisms was then investigated using the same technique.

In *Escherichia coli*, strain B, as suggested by LANNING AND COHEN⁴, ribose appears to be synthesized primarily from hexose by loss of carbon 1⁵. In the rat, however, the data indicate that the direct pathway from hexose is not the only mechanism for the synthesis of the pentose. When rats were given $\text{NaH}^{14}\text{CO}_3$ and ribose and glycogen were isolated from the liver, the distribution of ^{14}C in the two carbohydrates were dissimilar.

Rats (about 300 g each) were starved for 48 hours and given intraperitoneally a total of $1.4 \cdot 10^9$ c.p.m. per 100 g body weight of $\text{NaH}^{14}\text{CO}_3$ (1.23 mM) in equal doses at 0, 0.5, 1, 2, 2.5, 3.5 and 4 hours and by stomach tube 200 mg of glucose at 0 time. At 4.5 hours, the animals were sacrificed and the ribose (from ribonucleic acid) and the glycogen were isolated from the pooled livers by the methods previously described². After hydrolysis of the glycogen, the resulting glucose was purified by filter paper chromatography, diluted 97 fold with unlabeled glucose and recrystallized⁶. The pattern of radioactivity in the hexose was determined by fermentation with *Leuconostoc mesenteroides*^{6*}. The ribose was also purified by paper chromatography² and after a 450 fold dilution with carrier was degraded for analysis of ^{14}C by fermentation with *Lactobacillus pentosus*^{7**}. Radioactivity data for the individual carbons, obtained by gas phase counting⁸, are presented in specific activities relative to that determined for one of the carbons given a value of 100.

The relative specific activities found were as follows***:

Glycogen	C — C — C — C — C — CHO	Ribose	C — C — C — C — CHO
	0.3 2 100 94 2 1		2 5 100 49 22
	(100 = $2.53 \cdot 10^5$ c.p.m./mM carbon)		(100 = $1.16 \cdot 10^3$ c.p.m./mM carbon)

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I. A. BERNSTEIN

Department of Dermatology, School of Medicine
and the Institute of Industrial Health, University Hospital,
University of Michigan, Ann Arbor, Mich. (U.S.A.)

¹ For reviews of this subject see S. S. COHEN, in D. M. GREENBERG, *Chemical Pathways of Metabolism*, Academic Press, New York, Vol. I, page 173, 1954. and G. E. GLOCK, in E. CHARGAFF AND J. N. DAVIDSON, *The Nucleic Acids*, Academic Press, New York, Vol. II, page 248, 1955.

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* Specific activity determinations on the recrystallized product indicated $5.19 \cdot 10^5$ c.p.m./mM of glucose. Potassium gluconate prepared from this material contained $5.05 \cdot 10^5$ c.p.m./mM of compound. Fermentation data showed $5.08 \cdot 10^5$ c.p.m./mM of hexose.

** Potassium ribonate prepared from the diluted material contained 2,150 c.p.m./mM of compound. Fermentation data indicated a specific activity of 2,060 c.p.m./mM of pentose.

*** The specific activities shown for the carbons given a value of 100 represent the levels at which the determinations were actually made on the diluted samples.